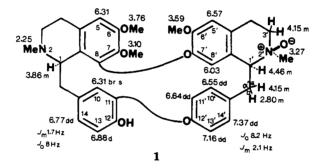
BERBAMINE 2'-β-N-OXIDE, A NEW BISBENZYLISOQUINOLINE FROM BERBERIS BRANDISIANA

S. Fazal Hussain, ¹ M. Tariq Siddiqui, ¹ Lajber Khan, ¹ Alan J. Freyer, Hélène Guinaudeau, ² and Maurice Shamma

Department of Chemistry, The Pennsylvania State University, University Park, Pennsylvania 16802

Berberis brandisiana Ahrendt (Berberidaceae) is a shrub that grows in northern Pakistan. A study of the alkaloidal contents of its aerial parts has resulted in the isolation of the new bisbenzylisoquinoline (+)-berbamine 2'- β -N-oxide (1), C₃₇N₄₀N₂O₇. sembles in it's general features the spectrum for berbamine (1). Notable divergences, however, were the chemical shifts of the 2'-N-methyl group and the aliphatic H- α' , H-1', and H-3'. It is known that the 2-N-methyl (or 2'-N-methyl) signal of a bisbenzylisoquino-



The mass spectrum of the alkaloid shows molecular ion m/z 624 of medium intensity (38%). This is accompanied by two strong peaks, m/z 608 (100%) and m/z 198 (97%). The m/z 608 peak is due to loss of an oxygen atom from the molecular ion. The m/z 198 peak corresponds to the doubly charged upper half of the m/z 608 species.

Inasmuch as the net loss of one oxygen in the mass spectrum is diagnostic of an N-oxide, our alkaloid was reduced with zinc in HCl to afford the known (+)-berbamine.

The 360 MHz (CDCl₃) nmr spectrum of berbamine $2'-\beta$ -N-oxide has been summarized around expression **1**. It re-

line N-oxide belonging to the same 8-7', 11-12' series as berbamine undergoes a downfield shift of 0.3-0.4 ppm when H-1 (or H-1') lies on the same side as the N-oxide oxygen, while a larger shift of 0.7-0.9 ppm is observed when H-1 (or H-1') is on the side opposite the N-oxide oxygen (1,2). In the specific instance of berbamine $2'-\beta$ -N-oxide (1), the 2'-N-methyl signal is at δ 3.27 whereas it falls at δ 2.57 in berbamine itself. The downfield shift of 0.7 ppm thus argues in favor of a β -N-oxide configuration in which H-1' and the Noxide oxygen are *anti* to each other.

Another criterion for establishing the stereochemistry at the N-oxide center concerns the chemical shift of H-1'. A downfield shift of ≈ 0.8 ppm has been observed for H-1 (or H-1') when syn to the N-oxide oxygen. The downfield shift is less pronounced when an *anti*-relationship obtains (1,2). In the present instance, H-1' appears at δ 4.46 whereas

¹Permanent address: PCSIR Laboratories, Peshawar, Pakistan.

²Permanent address: Faculté de Médecine et de Pharmacie, Université de Limoges, 87025 Limoges Cedex, France; or CNRS, UA 496, Centre d'Etudes Pharmaceutiques, 92290 Châtenay-Malabry, France.

the corresponding absorption in berbamine is at δ 3.85. This Δ δ 0.61 ppm thus corroborates the 2'- β -N-oxide arrangement.

To confirm the above stereochemical conclusion, an nmr nOe analysis was performed on berbamine $2'-\beta$ -N-oxide (1), the results of which are given in the Experimental. A significant feature of this analysis is the reciprocating enhancements between the 2'-N-methyl (δ 3.27) and the H-1' (δ 4.46) signals, so that these hydrogens must be in a *syn* relationship.

Known alkaloids presently found in B. brandisiana are the isoquinolone thalifoline, the tetrahydrobenzylisoquinoline (+)-reticuline, the aporphines (+)-apoglaziovine and (+)-isoboldine, the protoberberinium salts berberine and palmatine, and the bisbenzylisoquinolines (+)-isotentrandrine, (+)berbamine, and (+)-penduline. It should be noted that while the bisbenzylisoquinolines of Berberis species usually possess the R, S' configuration, as in (+)isotetrandrine, (+)-berbamine, and (+)-berbamine $2'-\beta$ -N-oxide (1), (+)penduline is unusual in that it incorporates the S, S' stereochemistry.

EXPERIMENTAL

PLANT MATERIAL.—The aerial parts of *B. brandisiana* (7 kg., dry) were collected in the Ushu valley of Swat in Northern Pakistan. A plant specimen was deposited in the National Herbarium in Islamabad.

EXTRACTION AND ISOLATION. --- Extraction of the powdered plant was with cold EtOH. The dried alcoholic extract was taken up in 5% HCl, filtered, and the filtrate basified with NH4OH. The mixture was extracted exhaustively with CHCl₃ to supply an alkaloidal fraction. This fraction (30 g), which also contained some lignans, was placed on a silica gel column (2 kg, 70-230 mesh) filled with CHCl₃. Elution was with CHCl₂ and then with CHCl₂ gradually enriched with MeOH. Major alkaloids found were berberine, palmatine, and (+)-berbamine. Minor alkaloids were thalifoline, (+)-reticuline, (+)apoglaziovine, (+)-isoboldine, (+)-isotetrandrine, and (+)-berbamine 2'-\u03b3-N-oxide (1) (12 mg). Known alkaloids were identified spectrally or by comparison with authenic samples.

(+)-BERBAMINE 2'-B-N-OXIDE (1). Amorphous, m/z 624 (M⁺, C₃₇H₄₀N₂O₇) (38), 608 (100), 594 (29), 396 (80), 382 (61), 206 (12), 198 (97), 175 (52); λ max (MeOH) 234 sh, 282 nm (log ϵ 4.55, 4.04); [α]D + 14° (c 0.08, MeOH). Significant nmr nOe's are 2'-NMe to H-1', 4%; H-1' to 2'-NMe, 3%; H-1' to H-8', 10%; 2'-NMe to H- α' a, 4%; H- α' b to H-14', 5%; H- α' b to H-8', 10%; H-1 to 2-NMe, 1.5%; 2-NMe to H-1, 7%; H-1 to H-10, 5%.

REDUCTION OF 1.—N-oxide 1 (4 mg) was stirred at room temperature for 2 h with powdered zinc (20 mg) and 10% HCl (5 ml). Workup provided (+)-berbamine (2 mg).

ACKNOWLEDGMENTS

This research was supported by NSF grants INT- 8209537 to M.S. and INT-8209796 to S.F.H.

LITERATURE CITED

- 1. H. Guinaudeau, A.J. Freyer and M. Shamma, Nat. Prod. Rep., (in press).
- 2. M. Lavault, A. Fournet, H. Guinaudeau and J. Bruneton, J. Chem. Res. (S), 248 (1985).

Received 9 December 1985